



Asymmetric synthesis of pre-protected α,α -disubstituted amino acids from *tert*-butanesulfinyl ketimines

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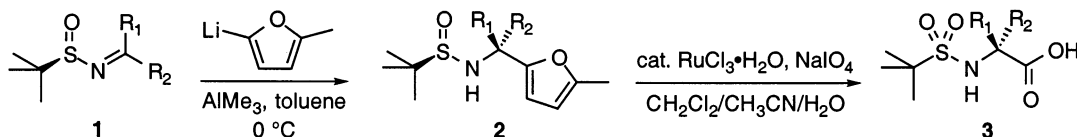
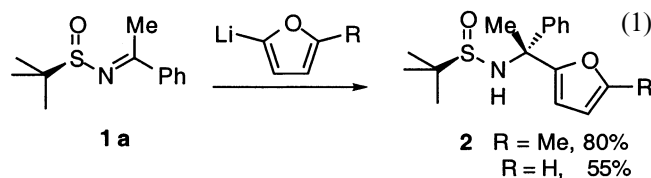
Abstract—A method for the asymmetric synthesis of pre-protected α,α -disubstituted amino acids is described. 5-Methylfuryllithium is added into sulfinyl ketimines **1** in the presence of AlMe_3 to afford the sulfinamides **2** in 75–97% yields and with diastereoselectivities ranging from 75:25 to 99:1. Subsequent oxidation with $\text{RuCl}_3/\text{NaIO}_4$ affords *tert*-butanesulfonyl (Bus)-protected α,α -disubstituted amino acids **3** in 62–69% yields. Bus-protected amino acids readily undergo amide bond formation, after which the Bus group can be removed with $\text{TfOH}/\text{CH}_2\text{Cl}_2$ to afford the free amine. © 2001 Elsevier Science Ltd. All rights reserved.

α,α -Disubstituted amino acids have displayed powerful effects on the conformation and the biological activity of molecules in which they are incorporated.¹ As a result, α,α -disubstituted amino acids have served as key components in numerous bioactive peptides, potential therapeutic agents, and natural products.² The most widely used methods for the asymmetric synthesis of α,α -disubstituted amino acids rely on diastereoselective enolate alkylation.² However, this approach prohibits the general use of poor $\text{S}_{\text{N}}2$ substrates, such as aryl halides or sterically demanding alkylating agents. Moreover, harsh conditions are often necessary to liberate the free acid from the alkylation product.

An alternative disconnection is the 1,2-addition of oxidizable carbanions to ketimines.³ We have recently reported the one-step preparation of sulfinyl ketimines **1** in high yields by condensation of enantiomerically pure *tert*-butanesulfinamide with ketones.⁴ Additions of organolithiums to **1** proceed cleanly and with high diastereoselectivities.⁵ Herein, we report the application of this methodology to the asymmetric synthesis of pre-protected α,α -disubstituted amino acids. Specifically, the high-yielding and diastereoselective 1,2-addition of 5-methylfuryllithium to **1** followed by oxidation

with $\text{RuCl}_3(\text{cat.})/\text{NaIO}_4$ affords *tert*-butanesulfonyl (Bus)-protected amino acids **3** in two steps from sulfinyl ketimine **1** (Scheme 1).⁶

We began by evaluating the 1,2-addition of oxidizable nucleophiles to sulfinyl ketimine **1a** (Eq. (1)). Ketimine **1a** was added to furyllithium and 5-methylfuryllithium under the optimal conditions we have previously reported, namely, with toluene as solvent and with 1 equiv. of AlMe_3 at -78°C .⁵ Unfortunately, due to the poor solubility of the furyllithium derivatives in toluene, poor conversions ($\leq 30\%$) were observed even when the reaction mixtures were allowed to warm to room temperature prior to quenching. By performing the reactions at 0°C a good yield (80%) and excellent diastereoselectivity (99:1 dr) was observed for the addition of 5-methylfuryllithium and a moderate yield (55%) and comparable diastereoselectivity was observed for the addition of furyllithium.



Scheme 1.

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Table 1. 1,2-Addition of 2-methylfuryllithium to sulfinyl ketimines **1** (Scheme 1)

Ketimine	R ¹	R ²	2 , Yield (%)	2 , dr	3 , Yield (%)
1a	Me	Ph	80	99:1 ^c	62 ^a
1b	Bu	Ph	87	97:3 ^d	67 ^b
1c	Me	<i>i</i> -Pr	97	94:6 ^d	63 ^a
1d	Bu	<i>i</i> -Pr	91	80:20 ^c	69 ^b
1e	Me	<i>i</i> -Bu	75	75:25 ^c	69 ^b

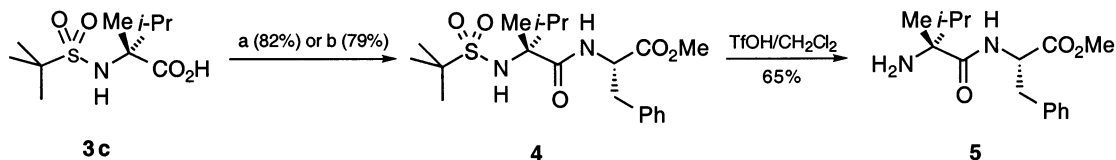
^a Absolute configuration determined by chemical correlation.⁸

^b Absolute configuration tentatively assigned on the basis of consistent diastereofacial selectivity observed for **1a**, **1c** and previously reported 1,2-addition reactions to sulfinyl ketimines.⁵

^c Diastereoselection determined by chiral HPLC analysis of the acetamide formed after sulfinyl cleavage.

^d Diastereoselection determined by chiral HPLC analysis of the benzamide formed after sulfinyl cleavage.

^e Diastereoselection determined by ¹H NMR.



Scheme 2. (a) HATU, Et₃N, DMF, then HCl·Phe-OMe; (b) i. SOCl₂, CH₂Cl₂, cat. DMF. ii. HCl·Phe-OMe, Na₂CO₃/NaHCO₃, CH₂Cl₂/H₂O.

The generality of the 1,2-addition of 5-methylfuran to *tert*-butanesulfinyl ketimines was then investigated (Table 1).⁷ Good to high yields were obtained for both dialkyl and aryl alkyl sulfinyl ketimines. The diastereoselection was highly dependent on the imine substitution. Thus, the diastereoselection was very high for imines with highly differentiated substituents (**1a–1c**), but dropped significantly when the substituents were of similar steric size (**1d–1e**).

Oxidation of sulfinamide **2a** with RuCl₃·H₂O (1 mol%) and NaIO₄ (15 equiv.) in 1:1:1.5 CH₂Cl₂/MeCN/H₂O⁶ went to completion in 2 h to afford the Bus-protected amino acid **3a** in 62% yield after acid/base extraction (Scheme 1, Table 1). When sulfinamide **2c** was oxidized under the same conditions, a 4:1 mixture of Bus-protected amino acid **3c** and the corresponding ketoacid was isolated after acid/base extraction. It was found that ketoacid formation could be partially suppressed (<10%) by oxidation with minimal acetonitrile (1:0.04:0.7 CH₂Cl₂/MeCN/H₂O), though the reaction was slower under these conditions (18 h). Acid/base extraction followed by recrystallization afforded analytically pure **3c** in 63% yield.^{9,10} Employing these modified conditions, Bus-protected amino acids **3b**, **3d** and **3e** were all obtained in good yields (Table 1). To demonstrate that no racemization occurs during the oxidation and Bus-protecting group removal steps, **3b** was converted to the (+)- and (–)-MTPA amides of α -butyl phenylglycine methyl ester by treatment with diazomethane, Bus deprotection with TfOH/CH₂Cl₂,⁶ and derivatization with (+)- or (–)-MTPACl, respectively. From starting carbinamine **2b** (97:3 dr), the MTPA amides were obtained in 97:3 dr as determined by HPLC analysis establishing that no degradation of stereochemical purity occurs during this synthesis sequence.

The utility of Bus-protected α,α -disubstituted amino acids as reagents for peptide synthesis was investigated by coupling amino acid **3c** with L-phenylalanine methyl ester (Scheme 2). Two coupling conditions were evaluated.¹¹ **3c** was activated in situ with HATU (1.3 equiv.) and Et₃N in DMF and coupled with 2 equiv. of HCl·Phe-OMe to deliver dipeptide **4** in 82% yield. Alternatively, **3c** could be converted to the acid chloride and then coupled with HCl·Phe-OMe in H₂O/Na₂CO₃-NaHCO₃/CH₂Cl₂¹² to deliver **4** in 79% yield. Significantly, despite the extremely hindered nature of the reactants, only 1.1 equiv. of the hydrochloride salt of phenylalanine methyl ester was required to achieve a high yield in this reaction. Subsequent cleavage of the Bus group from dipeptide **4** using TfOH/CH₂Cl₂⁶ afforded the free amine **5** in 65% yield.¹³

In conclusion, we have demonstrated the utility of *tert*-butanesulfinyl ketimines **1** for the asymmetric synthesis of pre-protected α,α -disubstituted amino acids in two steps. Carboxylic acid equivalent 5-methylfuryllithium can be added into both aryl alkyl and dialkyl ketimines in high yield and in moderate to high diastereoselectivities. The resulting sulfinamides are oxidized with RuCl₃(cat.)/NaIO₄ to afford Bus-protected α,α -disubstituted amino acids in 62–69% yields. Bus-protected α,α -disubstituted amino acids can be activated either as the acid chloride or with HATU to provide excellent coupling partners for amide bond formation, and should be versatile intermediates for the synthesis of hindered peptides and drugs.

Acknowledgements

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- General procedure for 5-methylfuryllithium additions to sulfinyl ketimines: (a) 5-methylfuryllithium was prepared by addition of 2.5 M *n*-BuLi in hexanes (5 equiv.) to a 5 M solution of 2-methylfuran (6 equiv.) in Et₂O at –10°C. The resulting solution was allowed to warm to room temperature and stirred overnight. (b) To a 1 M solution of sulfinyl ketimine (1 equiv.) in toluene at 0°C was slowly added a 1 M solution of AlMe₃ in toluene (1.1 equiv.). The resulting solution was stirred for 5 min before it was added over the course of 30 min to a 0°C solution of 5-methylfuryllithium diluted to 0.5 M with toluene. Stirring was continued at 0°C for 3–4 h before the mixture was allowed to warm slowly to room temperature. Saturated aq. Na₂SO₄ was added dropwise until gas was no longer evolved upon addition, and solid MgSO₄ was added. The slurry was stirred for 5 min before it was filtered through a Celite pad and the filter cake was rinsed with EtOAc. Chromatography of the residue left after concentration of the filtrate afforded sulfinamides **2a–e**. Product characterization for sulfinamide (*R*)-**2c**: IR: 1064, 1363, 1387, 1455 cm⁻¹. ¹H NMR (400 MHz) δ 0.76 (d, *J*=6.8, 3H), 0.93 (d, *J*=6.9, 3H), 1.17 (s, 9H), 1.43 (s, 3H), 2.19–2.25 (m, 4H), 3.37 (s, 1H), 5.83–5.84 (m, 1H), 6.11 (d, *J*=3.0, 1H). ¹³C NMR (101 MHz) δ 13.57, 16.79, 17.58, 19.12, 22.54, 36.81, 55.72, 59.81, 105.72, 107.28, 151.03, 156.30. Anal. calcd for C₁₄H₂₅NO₂S: C, 61.95; H, 9.28; N, 5.16. Found: C, 61.85; H, 9.27; N, 5.15.
- Compound **2a** was converted to the methyl ester of α -methyl-D-phenylglycine hydrochloride, and **2c** was converted to the methyl ester of α -methyl valine (see Ref. 10).
- Alternatively, the ketoacid could be scavenged from the product by overnight treatment of the crude material with 1 equiv. of polystyrene-*p*-toluenesulfonyl hydrazide resin in dichloroethane at 40°C.
- General procedure: RuCl₃·H₂O (1 mol%) was added to a mixture of NaIO₄ (15 equiv.) in CH₂Cl₂/MeCN/H₂O (1.0:0.04:0.7), and the mixture was stirred for 1 h. A solution of **2** in CH₂Cl₂ was added rapidly to the mixture via cannula. The final concentration of **2** was 0.03 M. Upon completion, as determined by TLC, the mixture was acidified to pH 1 with 1N NaHSO₄. The mixture was filtered through a plug of Celite and the organic layer was separated and washed once with brine followed by extraction with 5% K₂CO₃ (3×). The aqueous layer was acidified to pH 1 with solid NaHSO₄ and then extracted with EtOAc (3×). The combined organic portions were dried (Na₂SO₄) and concentrated. For amino acid **3c**, acid/base extraction was followed by recrystallization from toluene. Product characterization for amino acid (*R*)-**3c**: IR: 1125, 1305, 1710, 1730, 2979, 3261 cm⁻¹. ¹H NMR (400 MHz) δ 0.99 (d, *J*=6.8, 3H), 1.02 (d, *J*=6.9, 3H), 1.42 (s, 9H), 1.56 (s, 3H), 2.11 (m, 1H), 4.47 (s, 1H), 10.28 (br s, 1H). ¹³C NMR (101 MHz) δ 16.98, 17.08, 17.33, 24.35, 36.96, 60.15, 65.88, 179.39. Anal. calcd for C₁₀H₂₁NO₄S: C, 47.79; H, 8.42; N, 5.57. Found: C, 47.71; H, 8.30; N, 5.59. The product was determined to have the (*R*) configuration upon straightforward conversion to α -methyl-D-valine methyl ester. [α]_D²³ –13.0 (*c* 6.1, CHCl₃). Lit. for α -methyl-L-valine methyl ester: [α]_D²³ +13.5 (*c* 2.4, CHCl₃). Tabcheh, M.; El Achqar, A.; Pappalardo, L.; Roumestant, M.-L.; Viallefont, P. *Tetrahedron* **1991**, *47*, 4611–4618.
- General procedure for coupling of **3c** with Phe-OMe: Method A: HATU (1.3 equiv.) was added to a 0.4 M solution of amino acid **3c** (1 equiv.) and Et₃N (4 equiv.) in DMF. After stirring for 1 h, HCl-Phe-OMe (2 equiv.) was added. Upon completion, as determined by TLC, the mixture was diluted with 3:1 hexanes/CH₂Cl₂ and washed with 1N NaHSO₄ (3×). The organic layer was washed with 0.1 M K₂CO₃ (3×), dried (Na₂SO₄), and concentrated. Chromatography with 85:15 hexanes/EtOAc afforded dipeptide **4**. Method B: This procedure is a modification of the reaction conditions reported by Vedejs (Ref. 12). (a) To a 0.2 M solution of amino acid **3c** (1 equiv.) and SOCl₂ (3 equiv.) in CH₂Cl₂ was added a catalytic amount of DMF. After stirring for 2 h, the solution was concentrated, dissolved in dry toluene, and concentrated again to yield the acid chloride, which was used without further purification. (b) To a stirred solution of HCl-Phe-OMe, NaHCO₃ (3.2 equiv.), and Na₂CO₃ (2 equiv.) in 1:1 CH₂Cl₂:H₂O (0.1 M) at 0°C was added a 0.1 M solution of the acid chloride in dry CH₂Cl₂. After stirring for 1 h, the reaction mixture was diluted with hexanes (3:1 hexanes/CH₂Cl₂ final ratio). The aqueous layer was washed with 3:1 hexanes/CH₂Cl₂ (3×). The combined organic layers were washed once with 0.5% HCl and then brine. Chromatography with 85:15 hexanes/EtOAc afforded dipeptide **4** in 79% yield.
- N*-sulfonyl-protected amino acid chlorides are markedly more stable than *N*-carbamate-protected amino acids, which decompose readily to the considerably less reactive oxazolinones. For the synthesis and application of *N*-sulfonyl amino acid chlorides, see: Vedejs, E.; Lin, S.; Klapars, A.; Wang, J. *J. Am. Chem. Soc.* **1996**, *118*, 9796–9797.

13. Deprotection of dipeptide **4** [this procedure is a modification of the conditions developed by Weinreb (Ref. 6a)]: To a solution of **4** in anisole (20 equiv.) was added a 0.4 N solution of TFOH in CH₂Cl₂ (6 equiv.). After stirring for 24 h, the mixture was washed with saturated NaHCO₃ (3×). After drying (Na₂SO₄) and concentration, chromatography (hexanes/EtOAc) afforded amine **5**.